

# Bioenergy utilization in Sweden

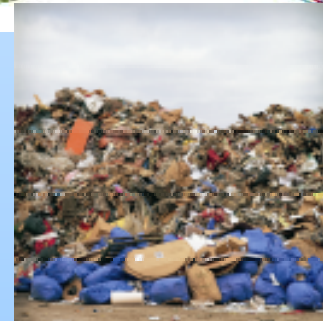
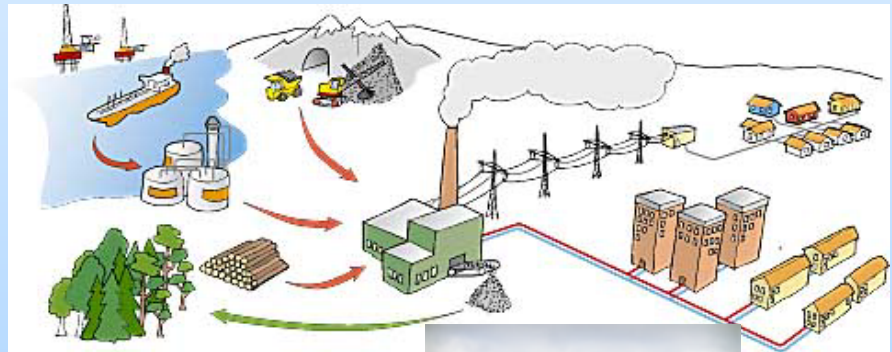
Claes Tullin

SP Swedish National Testing and  
Research Institute



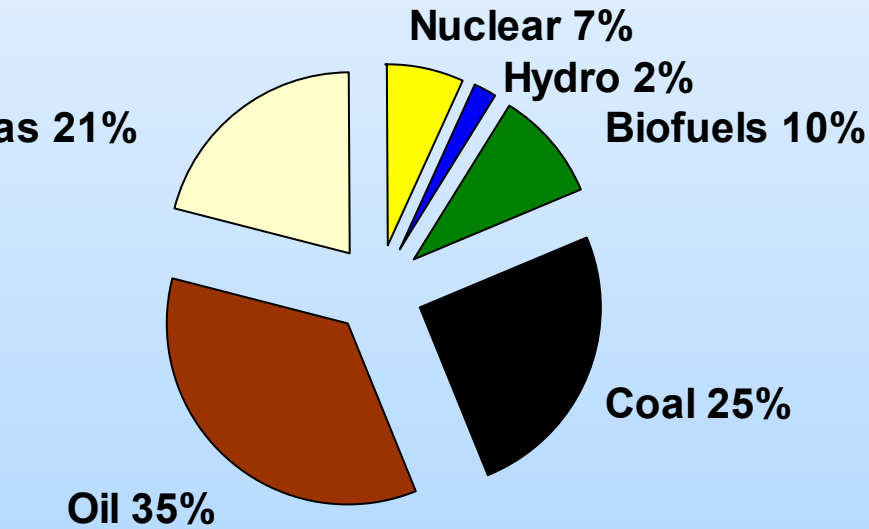
# Outline

- Introduction to biomass in Sweden
- Waste wood combustion
- Increasing market for refined fuels (pellets)

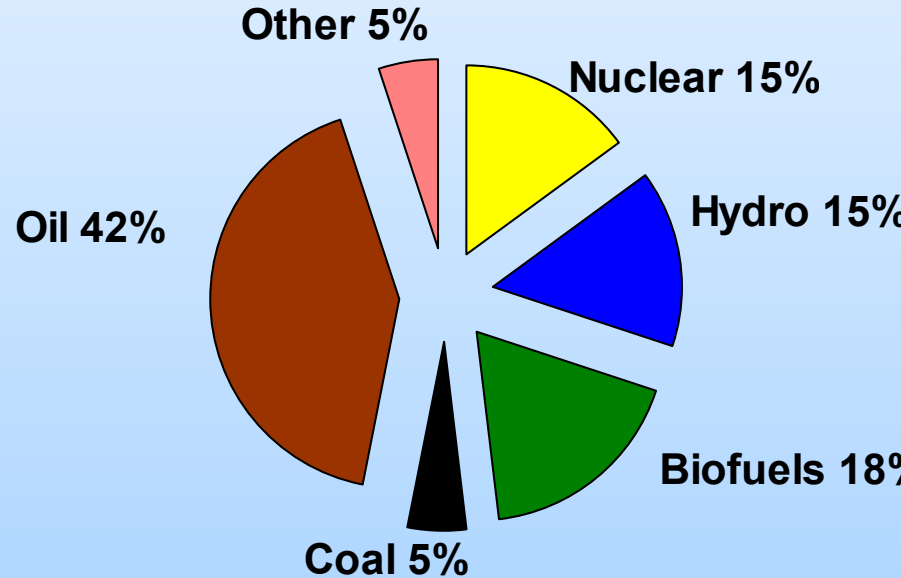


# Energy sources

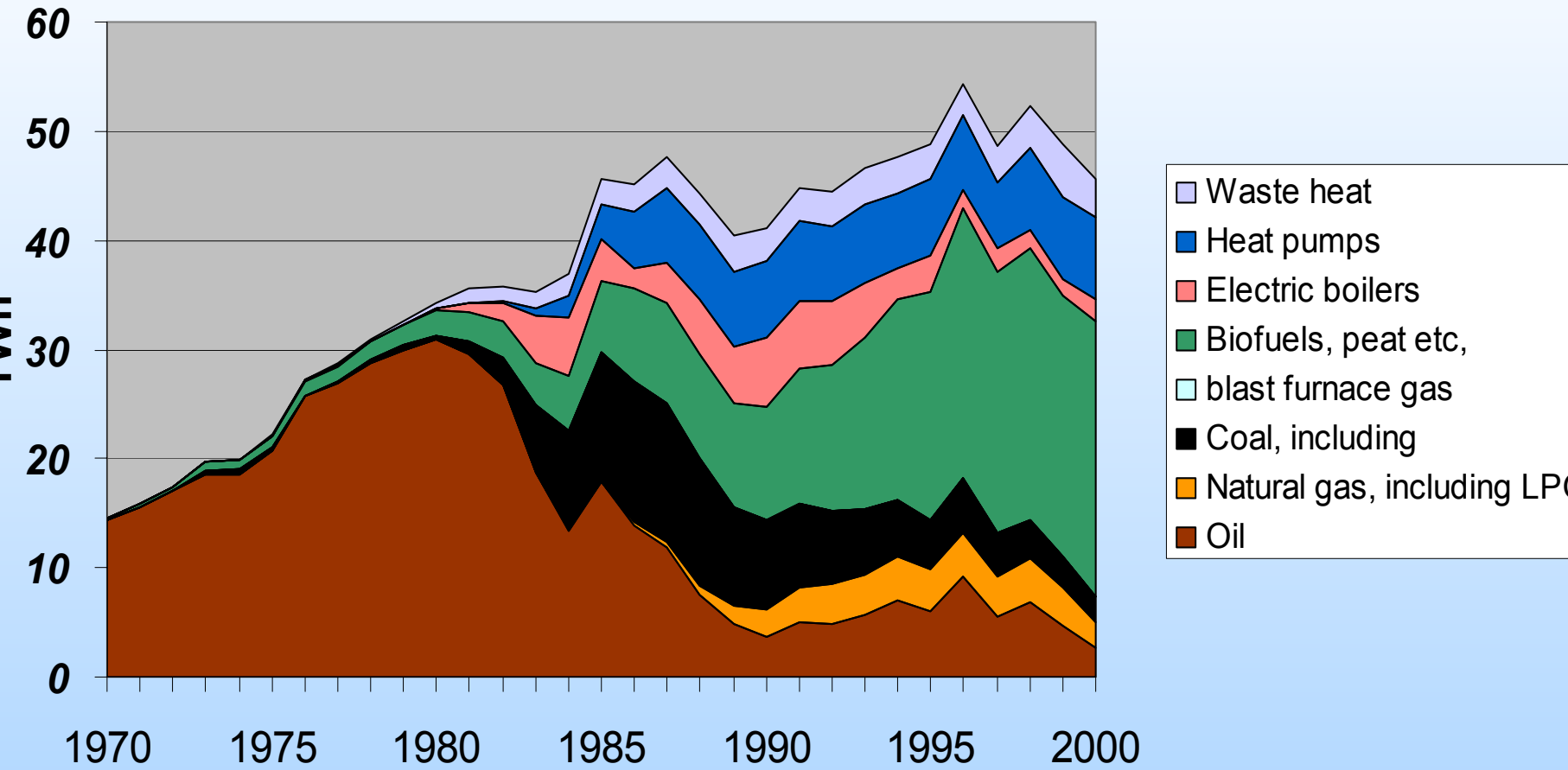
Global (100 000 TWh)



Sweden (484 TWh)



# Supply of district heating



# Driving mechanisms (1)

- Emissions of fossil CO<sub>2</sub> do enhance the greenhouse effect = a political truth!! => CO<sub>2</sub> reductions important!
  - Tax on CO<sub>2</sub> from fossil fuels
- Ban on landfilling of combustible wastes (from jan. 2002)
- Closing of nuclear reactors
- Power shortage

# Increasing use of Bioenergy

1970 – 2000:	+ 2 TWh/year
1980 – 2000:	+ 2.5 TWh/year
1990 – 2000:	+ 3.1 TWh/year
2000 – 2010:	+ ?? TWh/year

# District heating

- Forest residues
- Briquettes, pellets
- Sorted wastes
  - Waste wood
  - RDF
  - Sludge
- MSW

## 2. Dedicated waste wood combustion

- Contaminants due to treatments
  - CCA (Cu, Cr, As)
  - Paints (Zn, Pb, Cd, ..)
  - Boards (N, Cl, S)
- Increased fouling/corrosion
- Emissions (EU incineration directive)
  - Metals and HCl



# Waste wood project

## Swedish Thermal Engineering Research Institute

- Literature data ash chemistry
- Detailed fuel analysis
- Detailed field measurements
- Numerical simulation
- Laboratory studies on Zn/Pb chemistry
- Two case studies:
  - Grate combustion (vibrating grate; 117 MW<sub>th</sub>)
  - Fluid bed (BFB; 98 MW<sub>th</sub>)



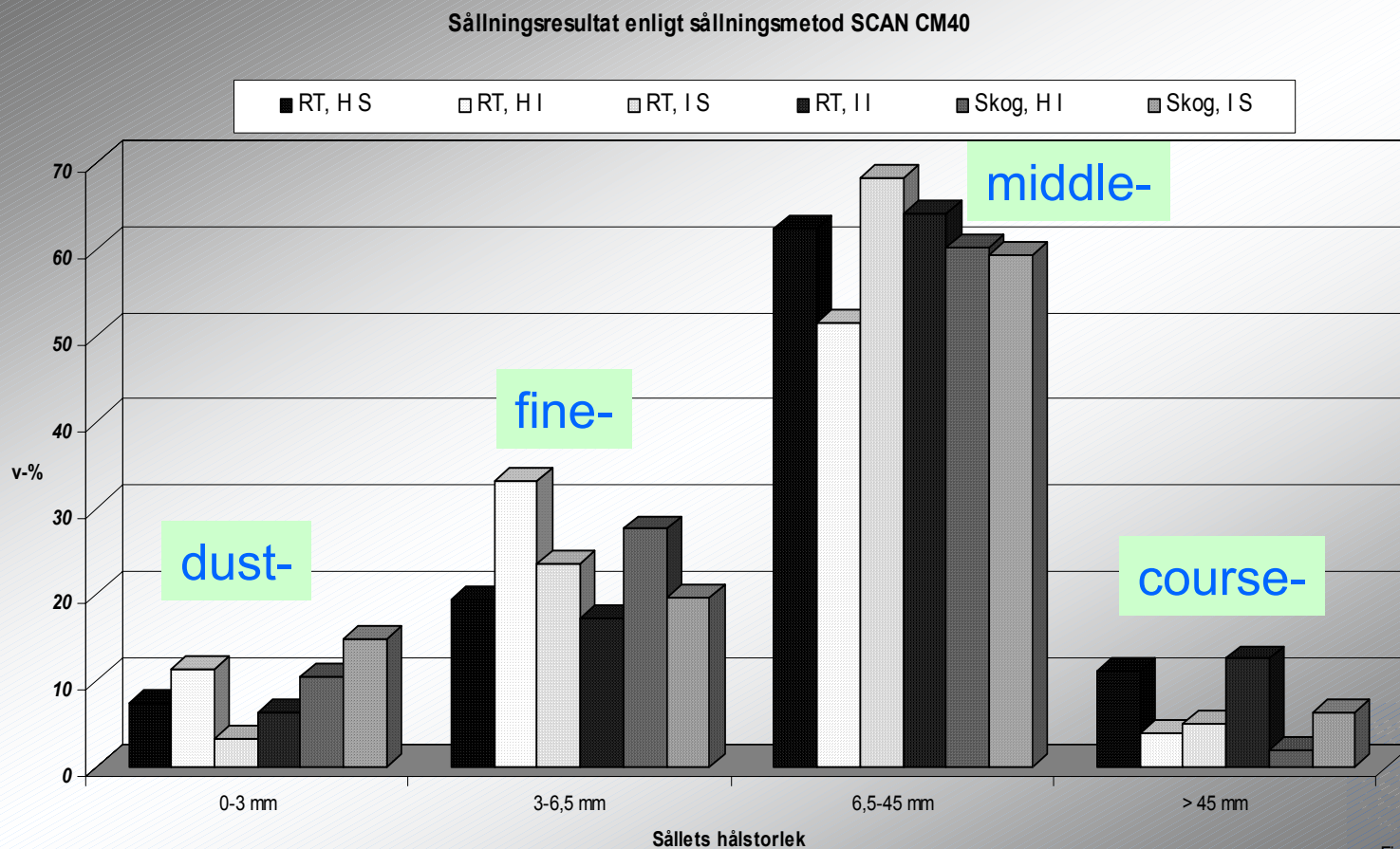
# Detailed fuel analysis

- Characterisation of fuel fractions
  - Painted, boards, plastics, metals
- Size distribution
- Chemical analyses

# Laboratory sieve => 5 fractions



# Sieving result



Figur 1

# Sieving result

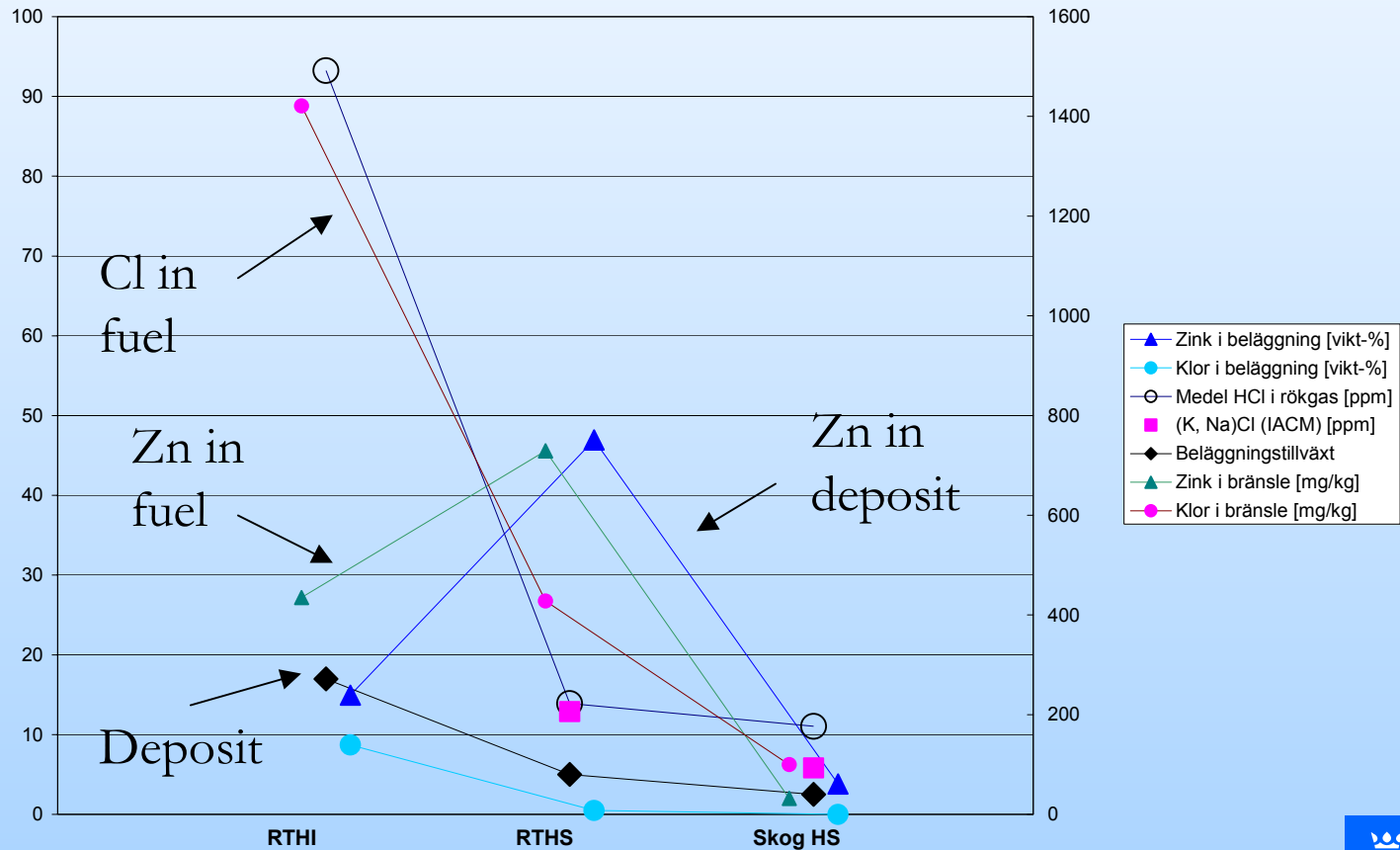
## Fines + dust = 30 w-%

- 60 % of the deposit related compounds (Zn, Pb, Na, K)
- 40 % for Cl

## Dust = 7 w-%

- 40 % of deposit related compounds (Zn, Pb, Na, K)
- 10 % för Cl

# Fuel quality – deposit formation





# Deposit probe test

50% biomix/50% recycled wood,  
12 h exposure,  $t_{\text{ring}} 500 \text{ }^{\circ}\text{C}$



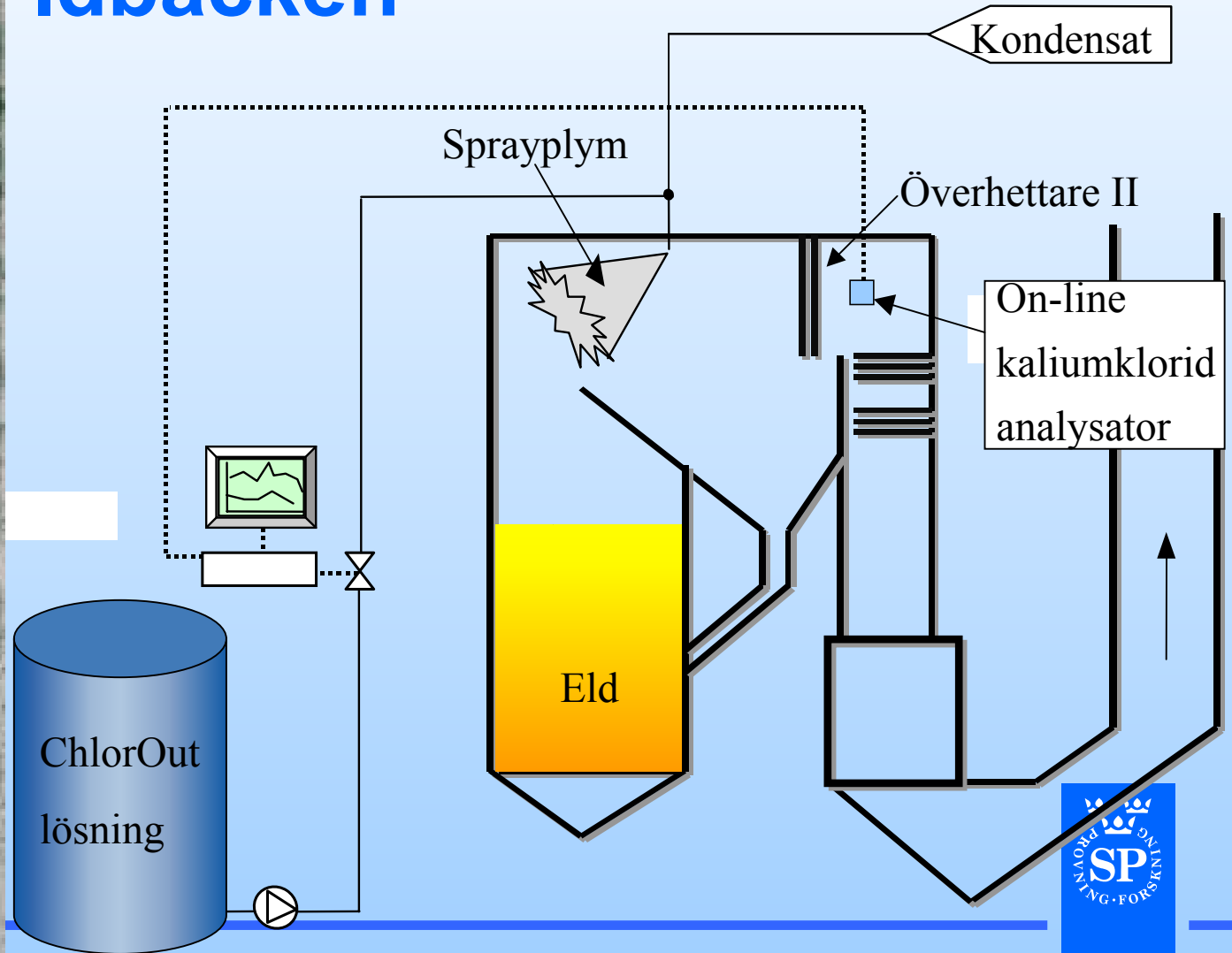
**Without ChlorOut;**  
chloride conc ~25%  
grow rate 21 g/m<sup>2</sup>/h



**With ChlorOut;**  
chloride conc <0.2%  
grow rate 6 g/m<sup>2</sup>/h

# ChlorOut installation

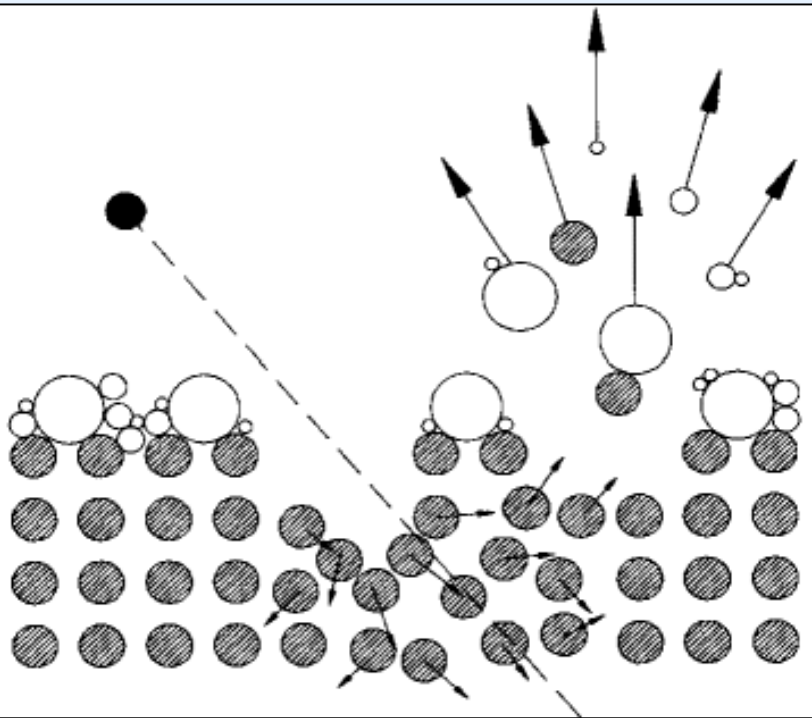
## Idbäcken





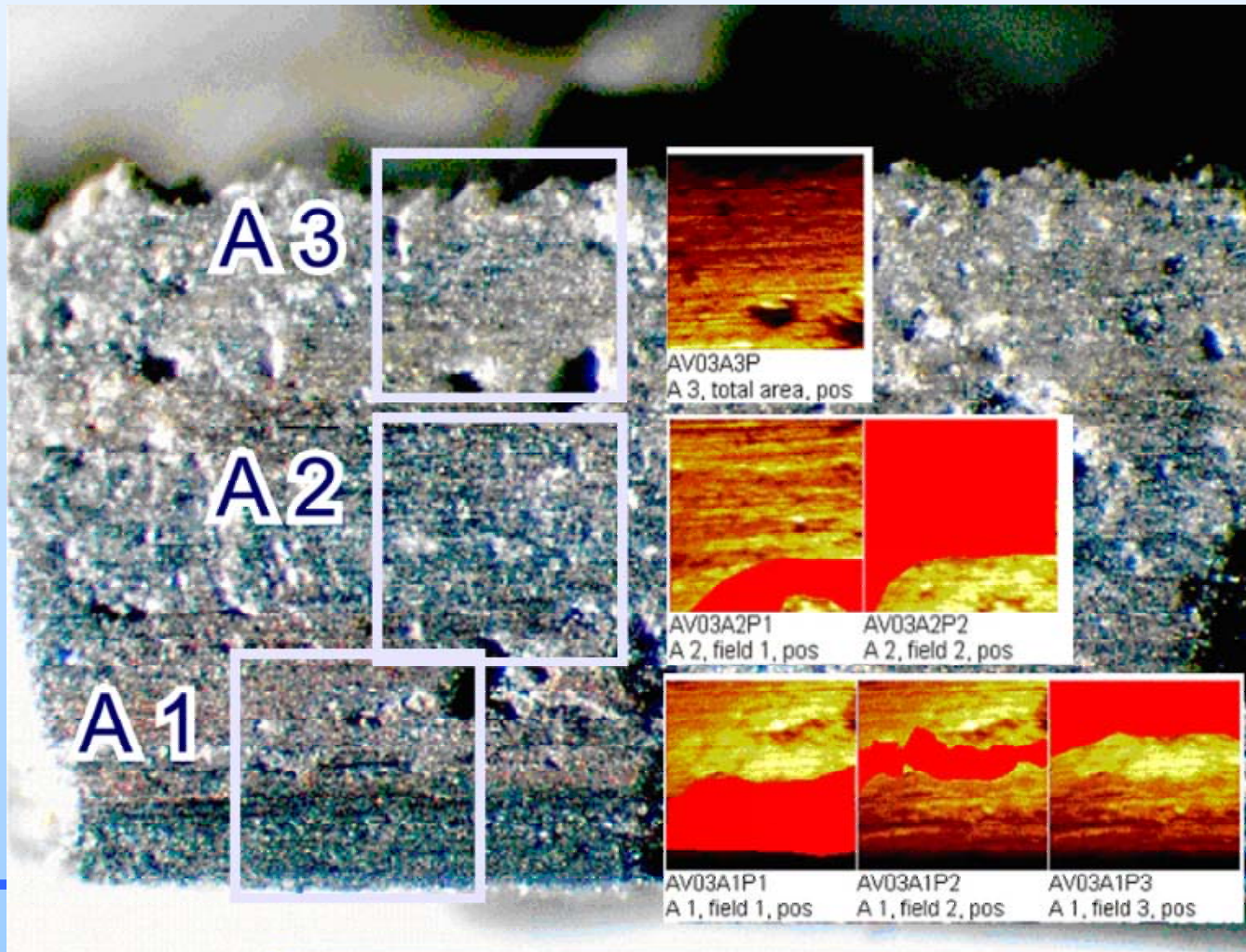
# TOF-SIMS - Time-of-Flight Secondary Ion Mass Spectrometry

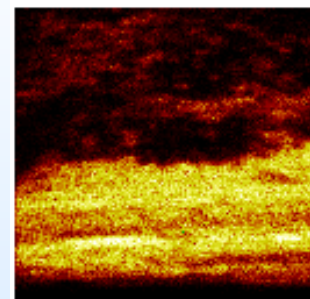
A high energy ion hits the surface and gives rise to a cascade of secondary ions from the outermost 1-2 atomic layers of the sample surface. The secondary ions are detected by a mass spectrometer.



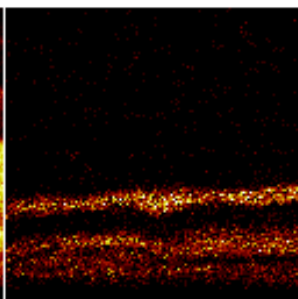
Schematic picture of  
the SIMS process.

# TOF-SIMS analysis of SH-deposit

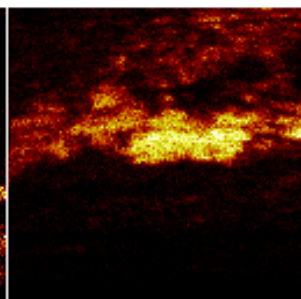




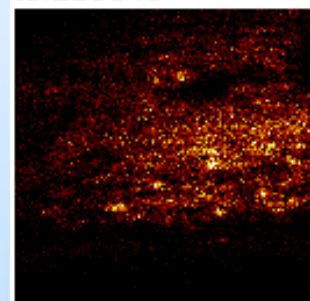
Fe  
tc:228846



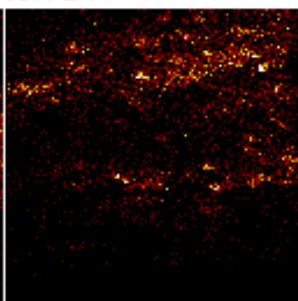
Cr  
tc:7247



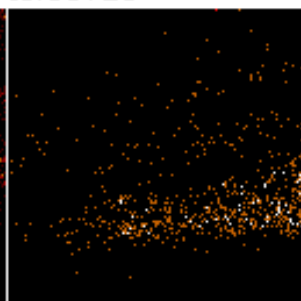
CaOH  
tc:85726



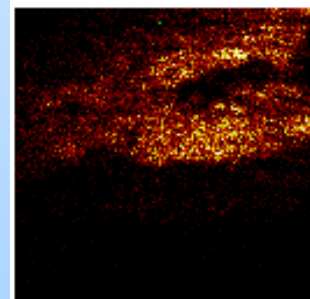
K<sub>2</sub>Cl (KCl)  
tc:15272



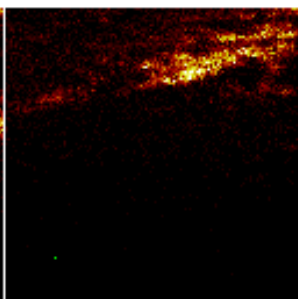
Na<sub>2</sub>Cl (NaCl)  
tc:4852



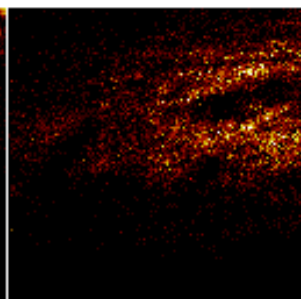
ZnCl<sub>3</sub> (ZnCl<sub>2</sub>)  
tc:689



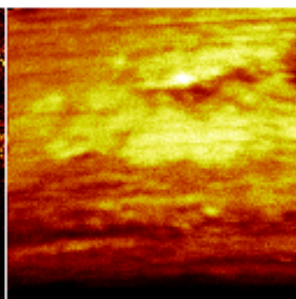
K<sub>2</sub>O (K<sub>2</sub>SO<sub>4</sub>)  
tc:16502



Na<sub>2</sub>O (Na<sub>2</sub>SO<sub>4</sub>)  
tc:10644

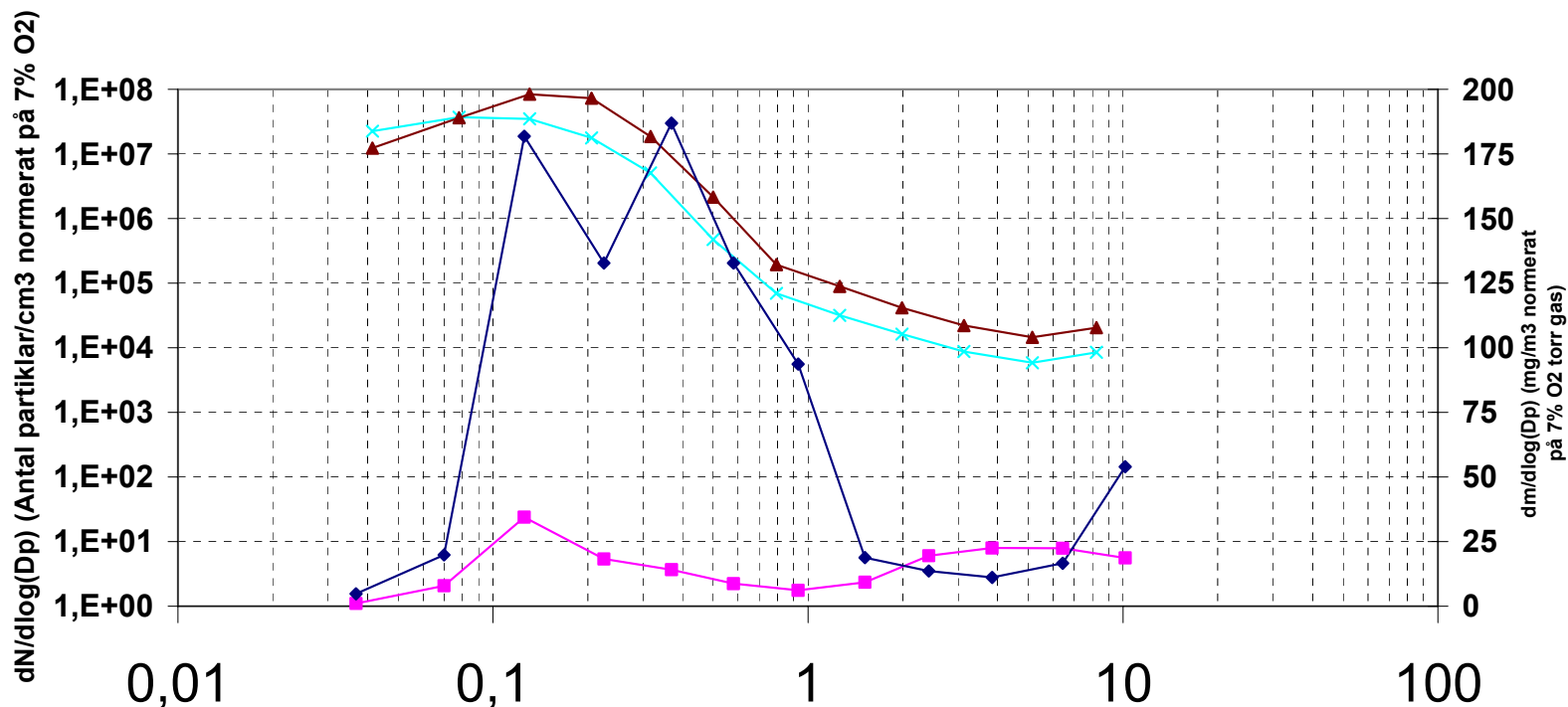


<sup>64</sup>ZnO<sub>2</sub> (ZnO)  
tc:7823



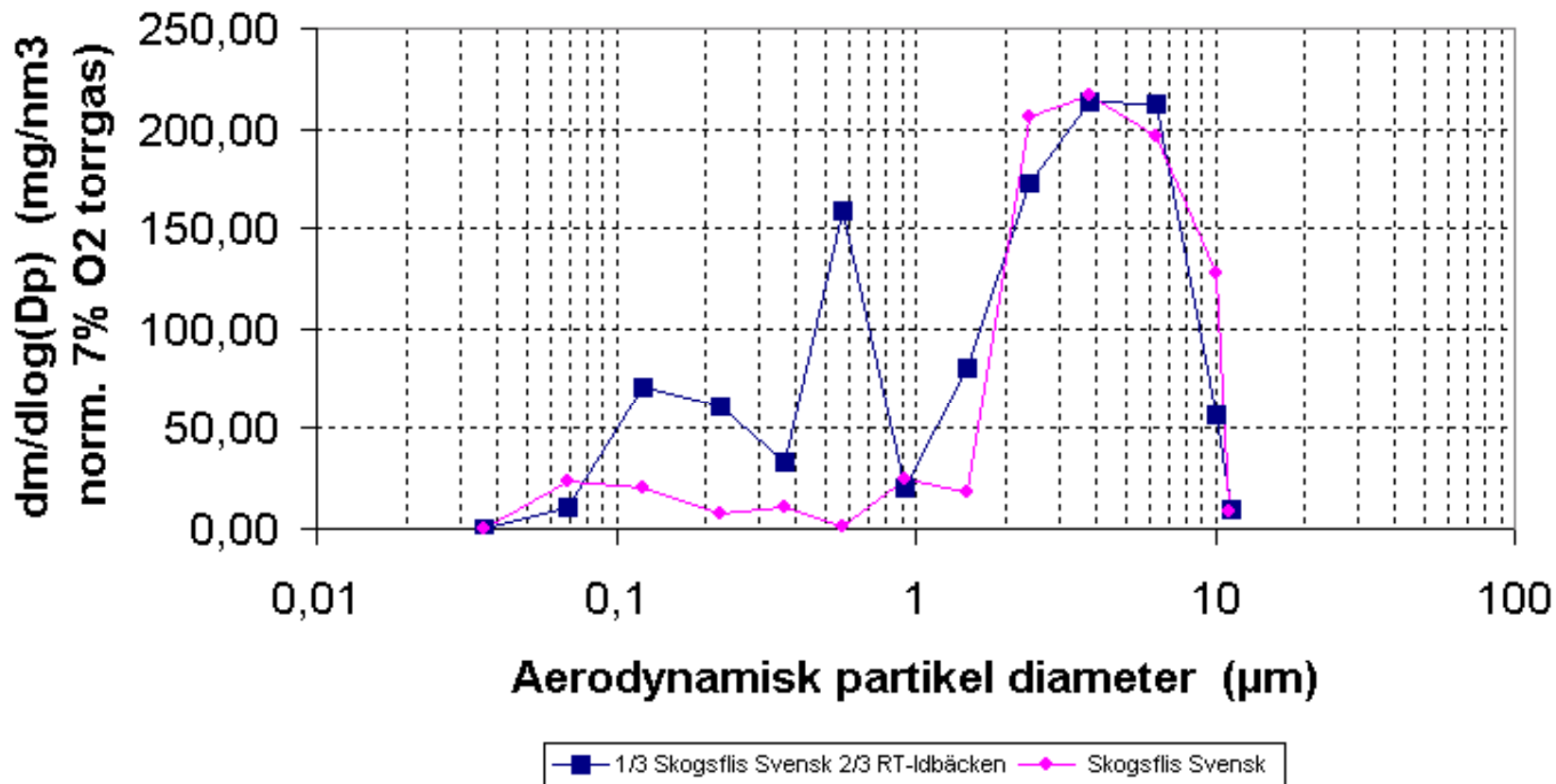
total ion  
tc:2897932

# Aerosol characteristics (grate)

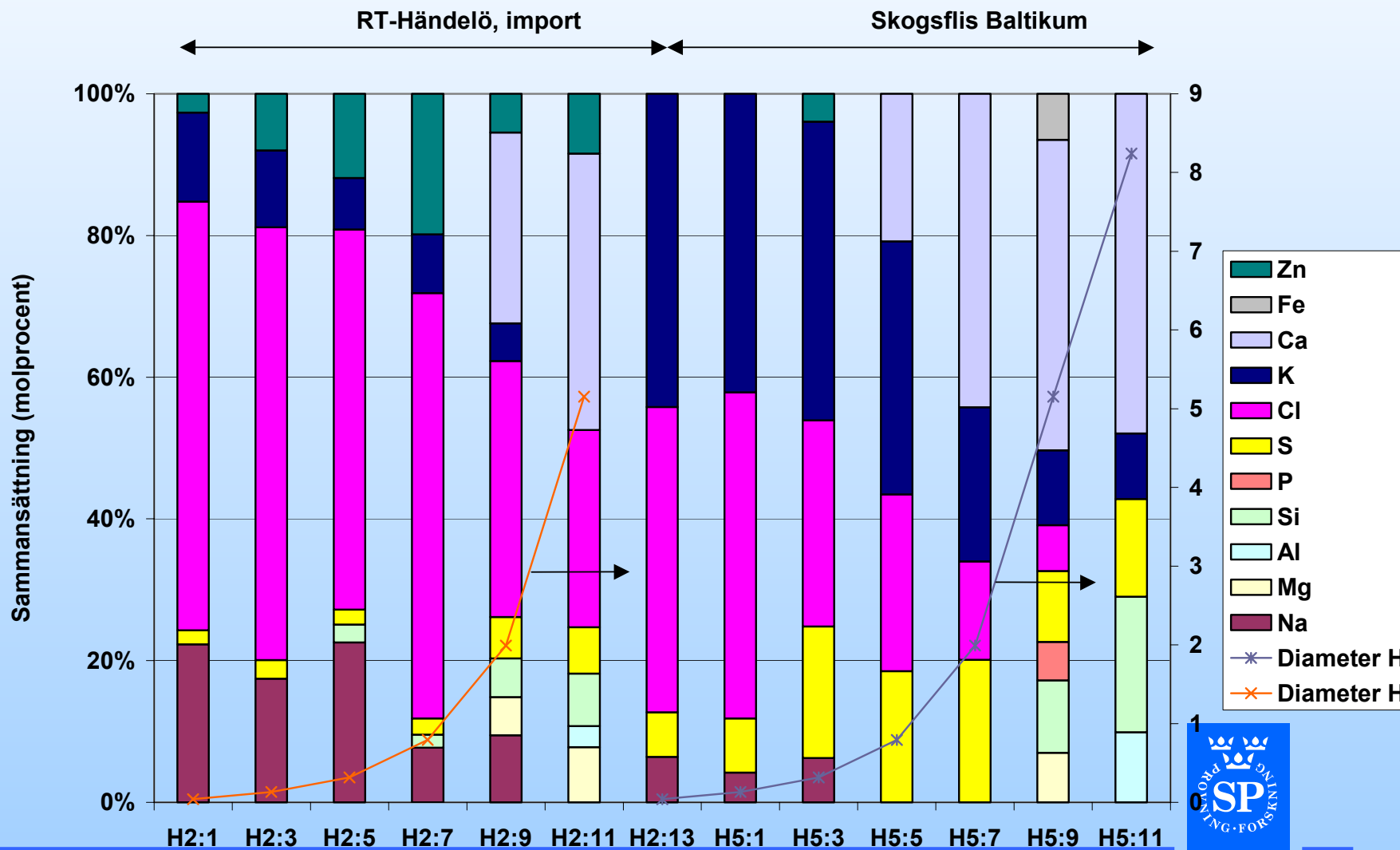


—x— Partikelkonc; Skog H I —▲— Partikelkonc; RT H I —■— Masskonc; Skog H I —◆— Masskonc; RT H I

# Aerosol characteristics (BFB)



# Chem analysis of PM fractions



# Conclusions: deposit formation

- In comparison with forest residues, combustion of waste wood =>
  - Higher rate of deposit formation
  - More Zn, Pb, K, Cl in the deposits
  - Increased corrosion
- Less Zn in FB deposits



# ZnO thermally stable....

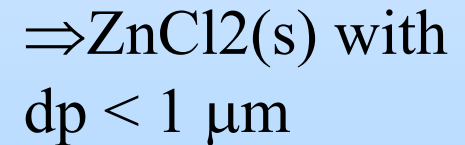
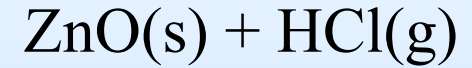
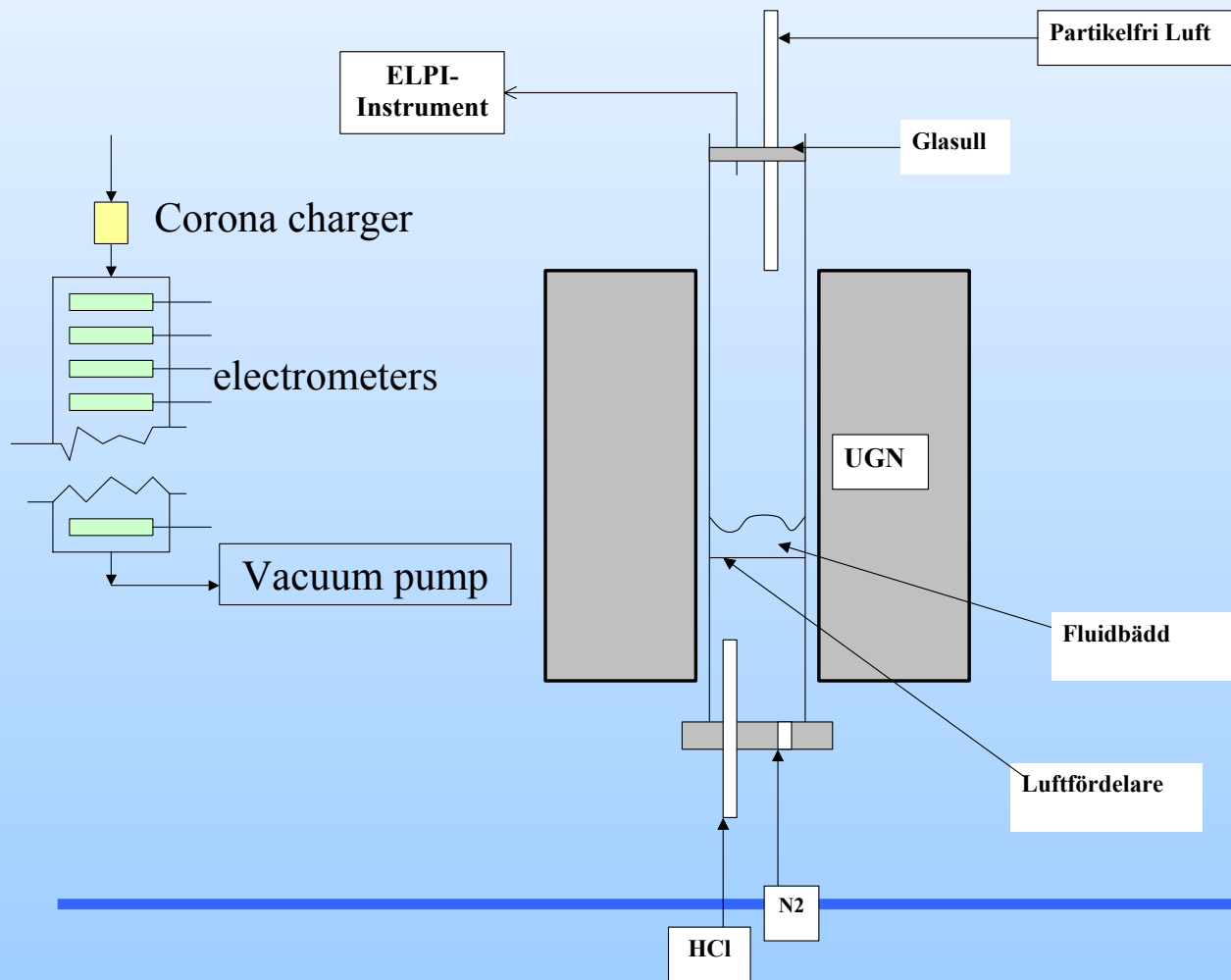
... but Zn still found in deposits and submicron particles:

Transport mechanisms from fuel bed to heat exchangers:

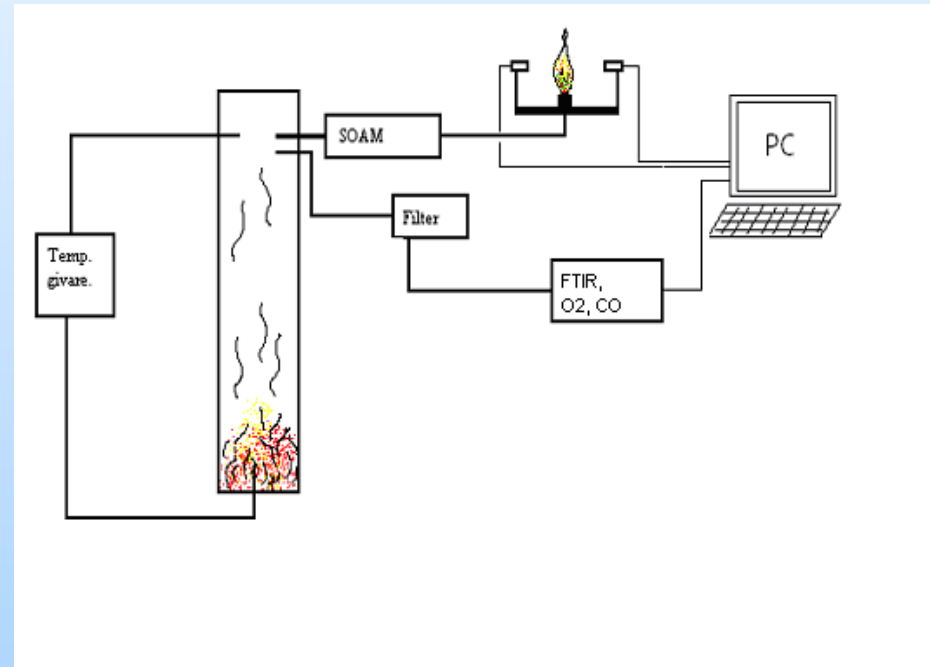
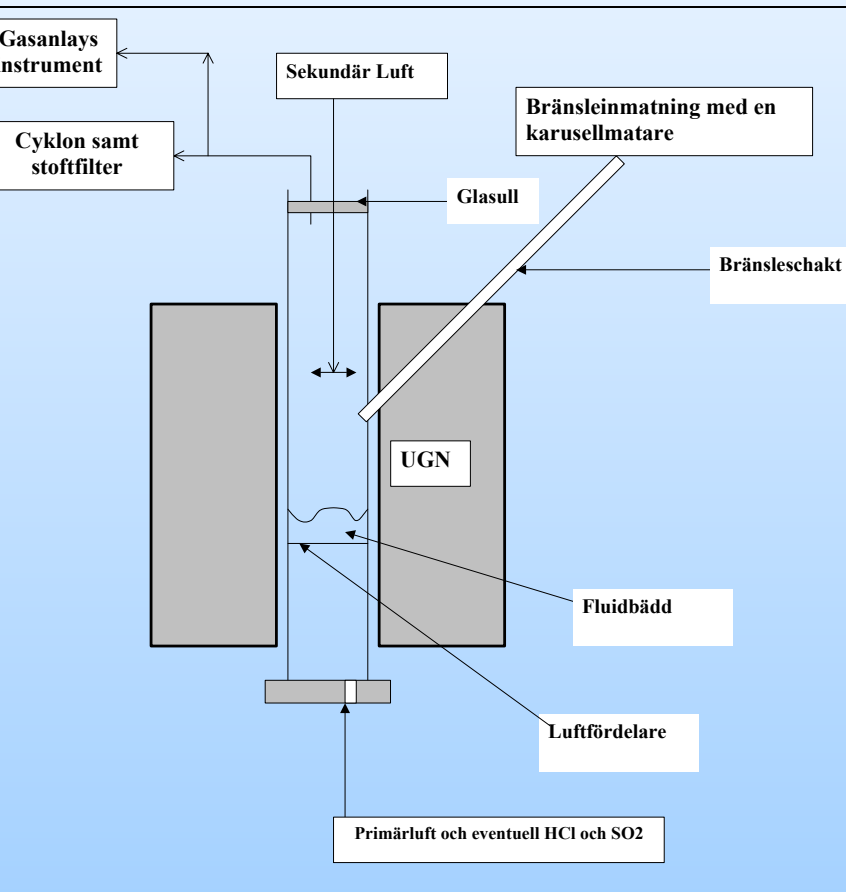
- ZnO pigment particles as such
- ZnO reduced to Zn which evaporates
- $\text{ZnO} + \text{HCl} \Rightarrow \text{ZnCl}_2$  which evaporates



# Experimental set-up



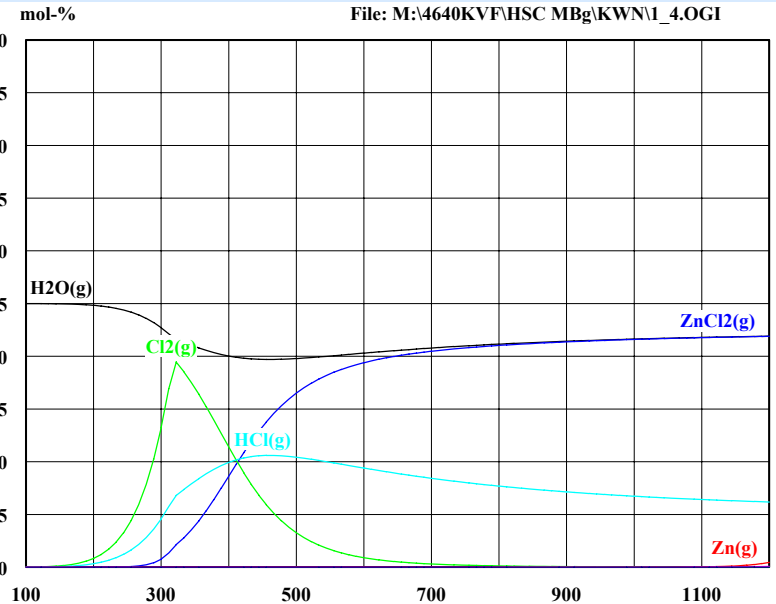
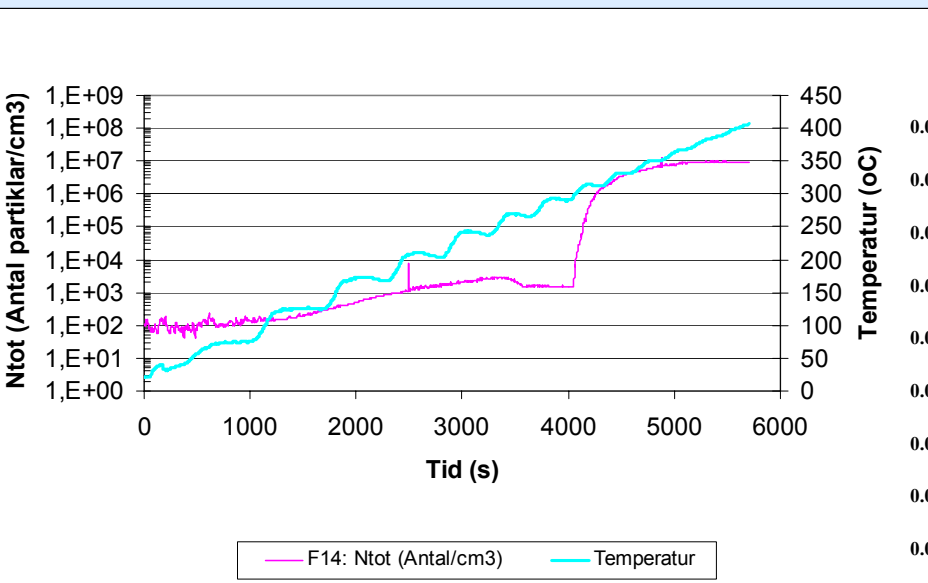
# Combustion tests with (1) waste fuel samples and (2) doped reference fuels



# Results - 1

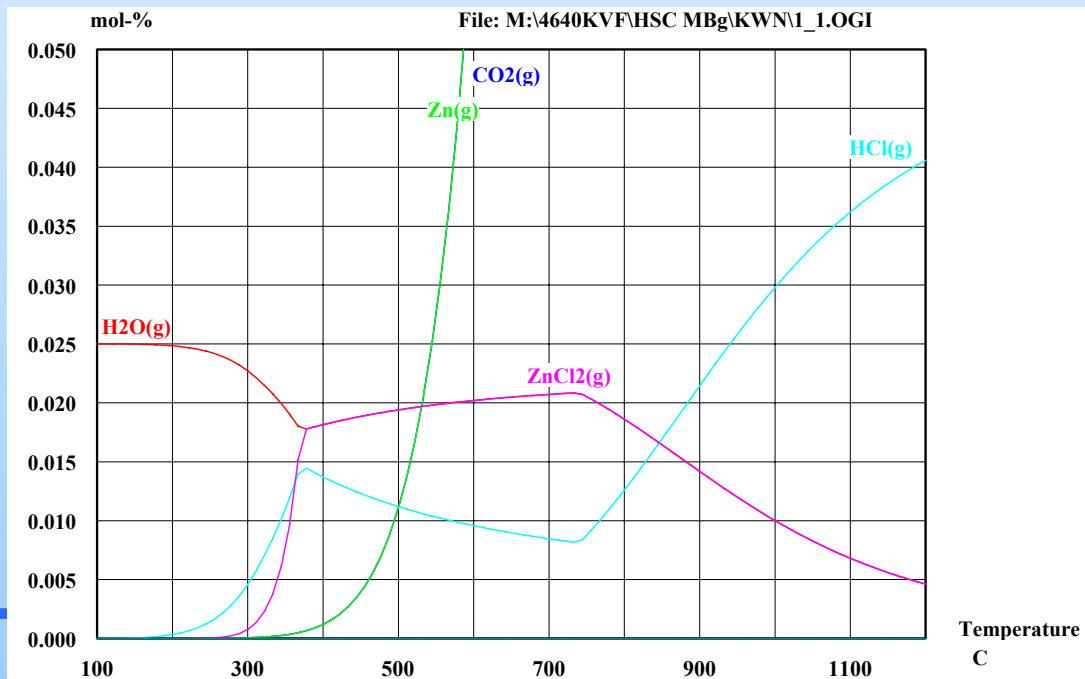
ZnO + 500 ppm HCl

Thermodynamic predictions



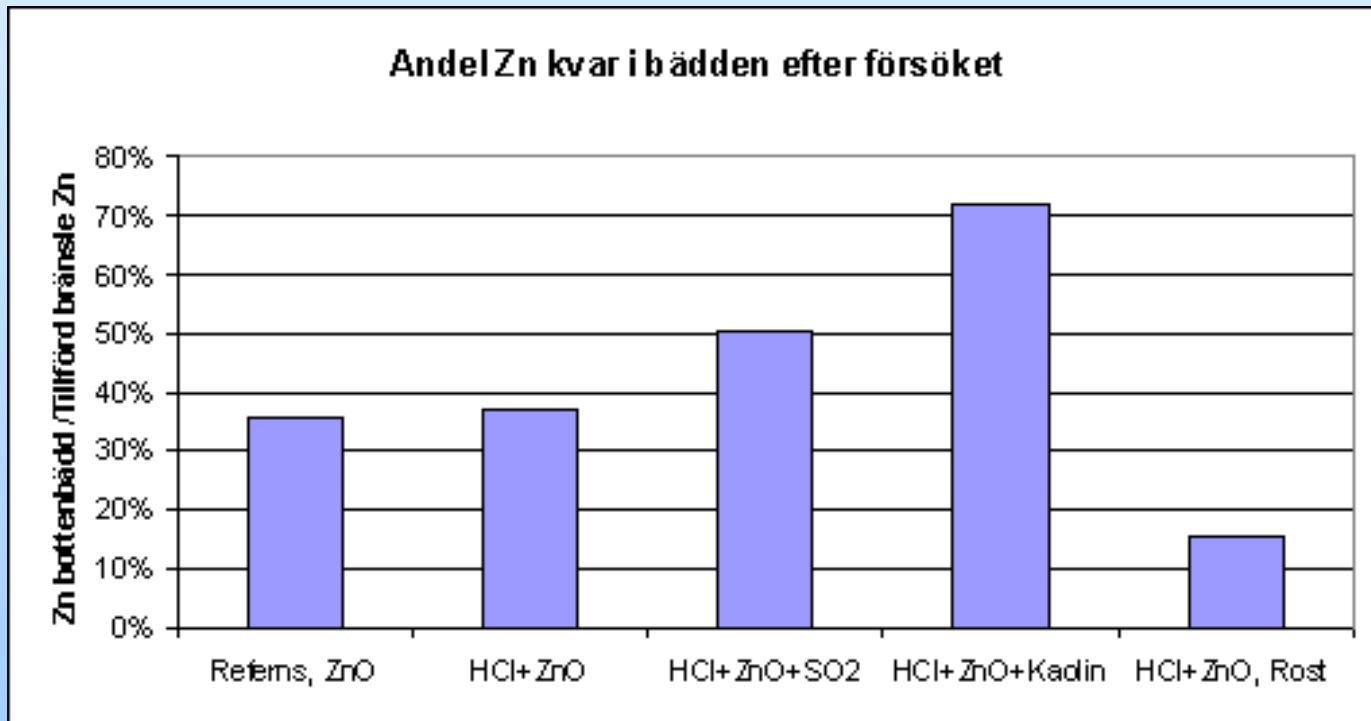
# Results: Formation of Zn

- Thermodynamic calculation predict Zn(g) formation but....
- this cannot be confirmed in experiments (CO, T = 850-1050 °C)  
=> kinetic limitations!?

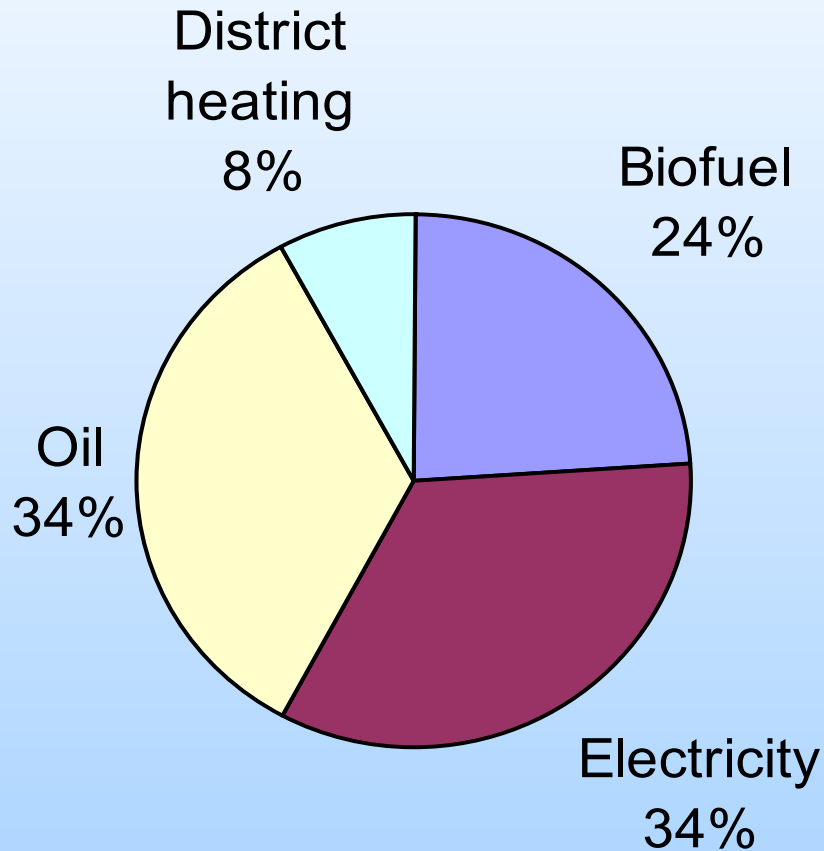


# Continuous combustion of painted wood -Additives reduce evaporation

- Addition of SO<sub>2</sub> and Kaolin reduce the vaporation of Zn
- Large difference between fluid bed and grate combustion conditions



# 3. Domestic heating in Sweden

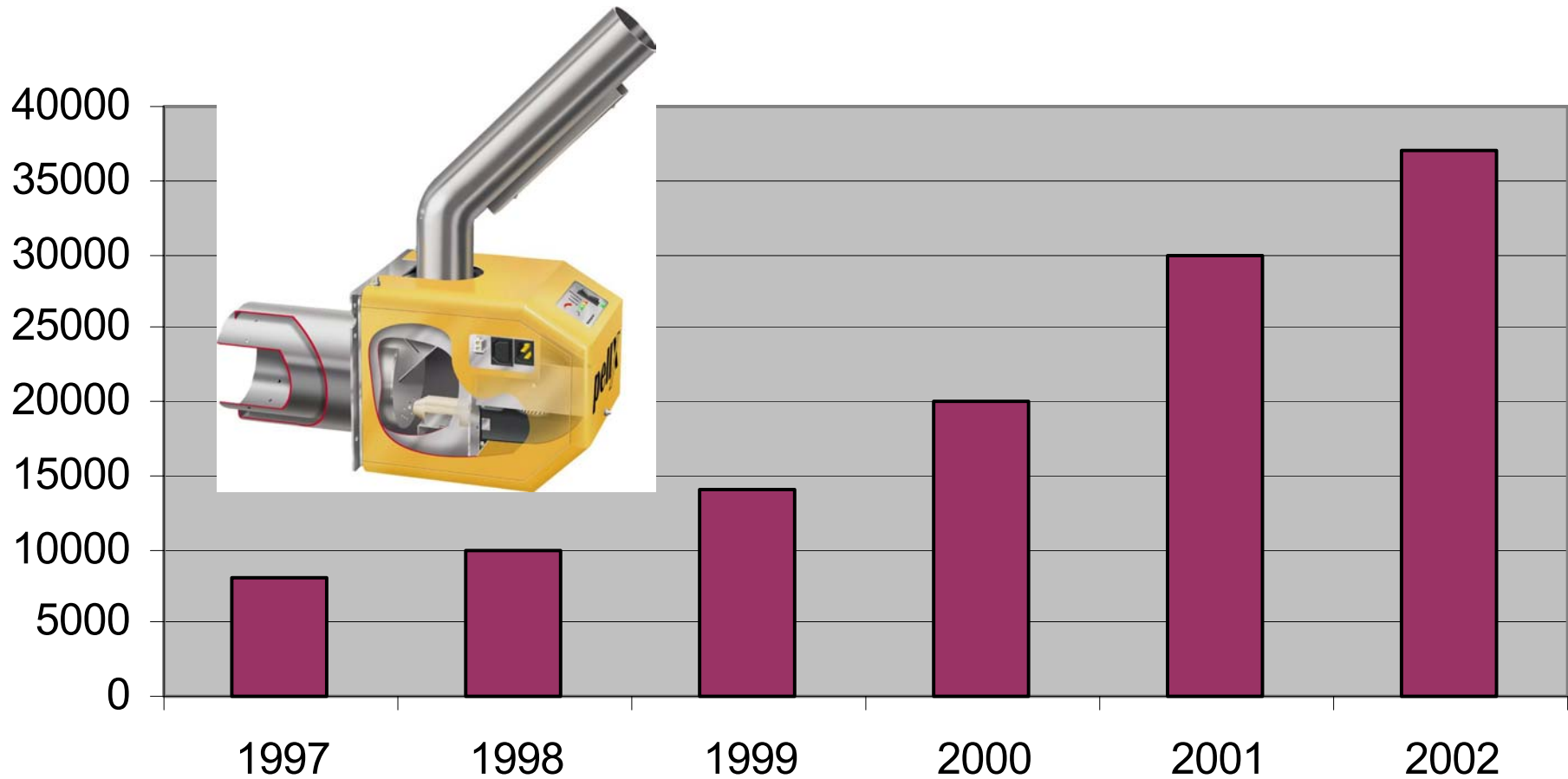


# Emissions from different domestic combustion equipment per net kWh

Type of combustion	Tar, mg	VOC, mg	NOx, mg	Particles,
Trad. wood boiler	2 500	10 000	350	1 800
Modern wood boiler	30	300	520	80
Modern wood stove	50	700	n.a.	80
Pellet boiler	20	160	<270	160
Pellet stove	20	120	<270	160

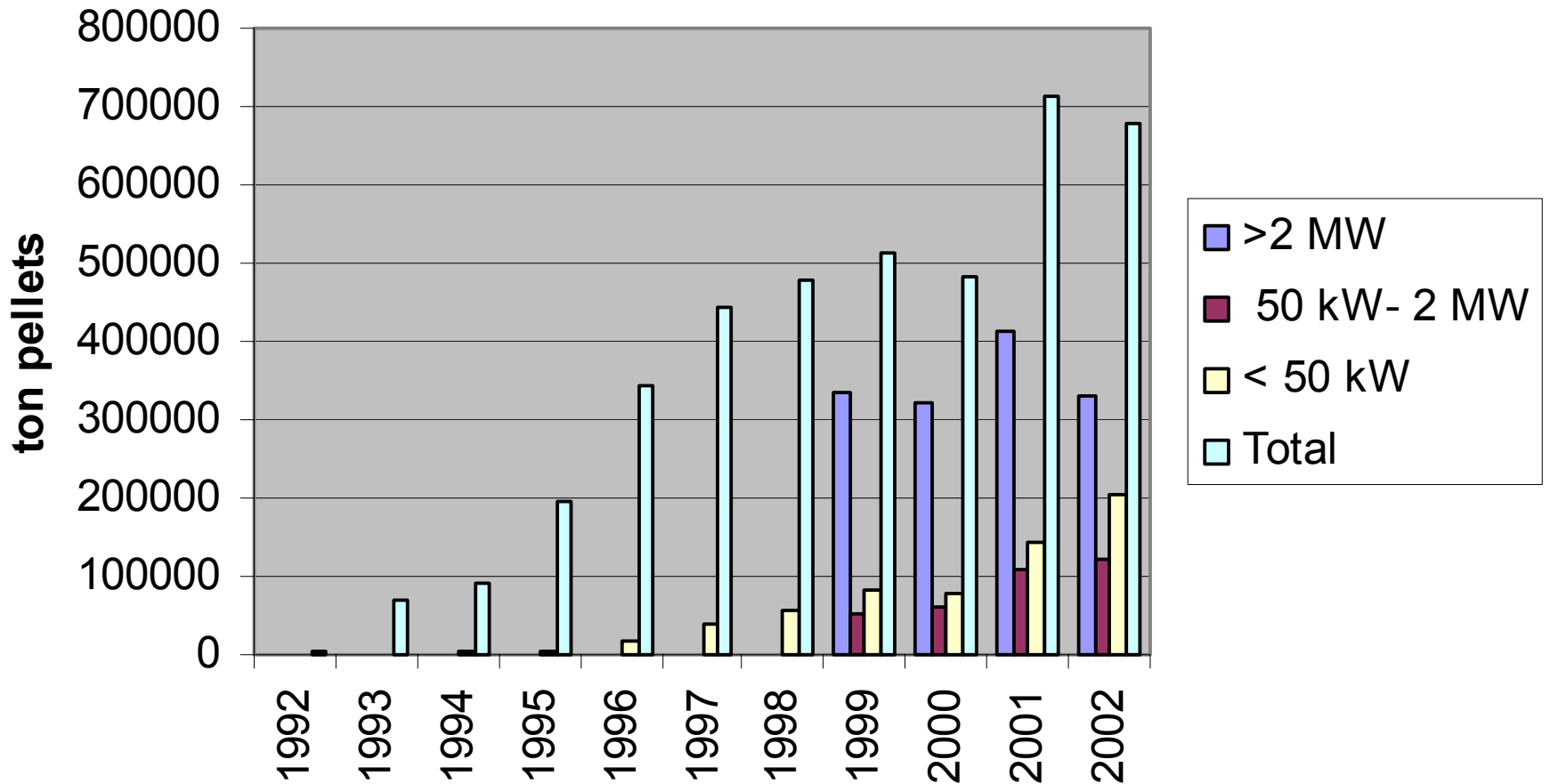
# Pellet burners

## Installed Pellet Burners





# Delivered pellets (Sweden)

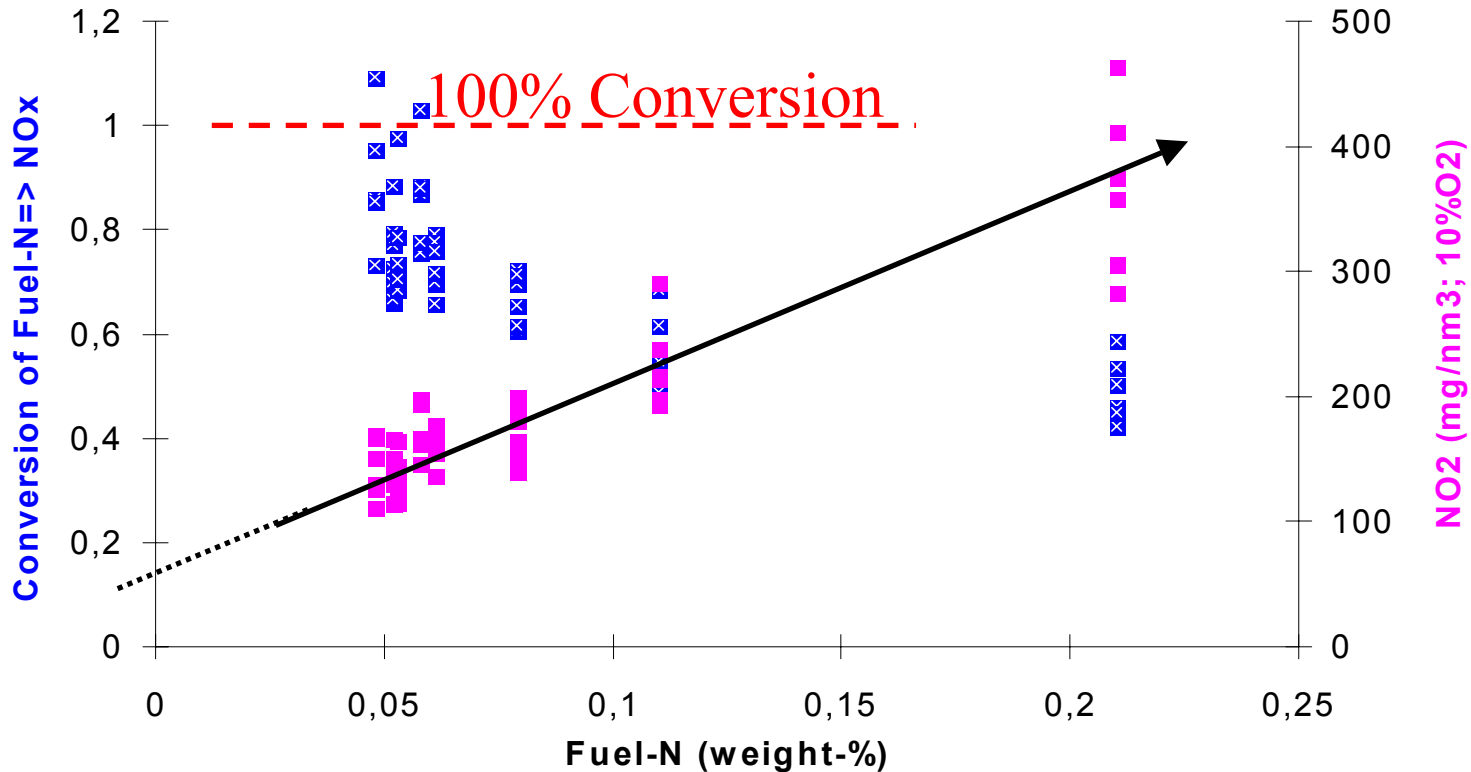


# Issues in Pellet Combustion

- Environmental problems?
  - NOx
  - Particulates
    - If replacing oil burners/electricity
- Technical issues
  - Pellet quality
  - Maintenance
  - Operator independent
  - Installation
- Economy/Market
  - Supply
- Safety

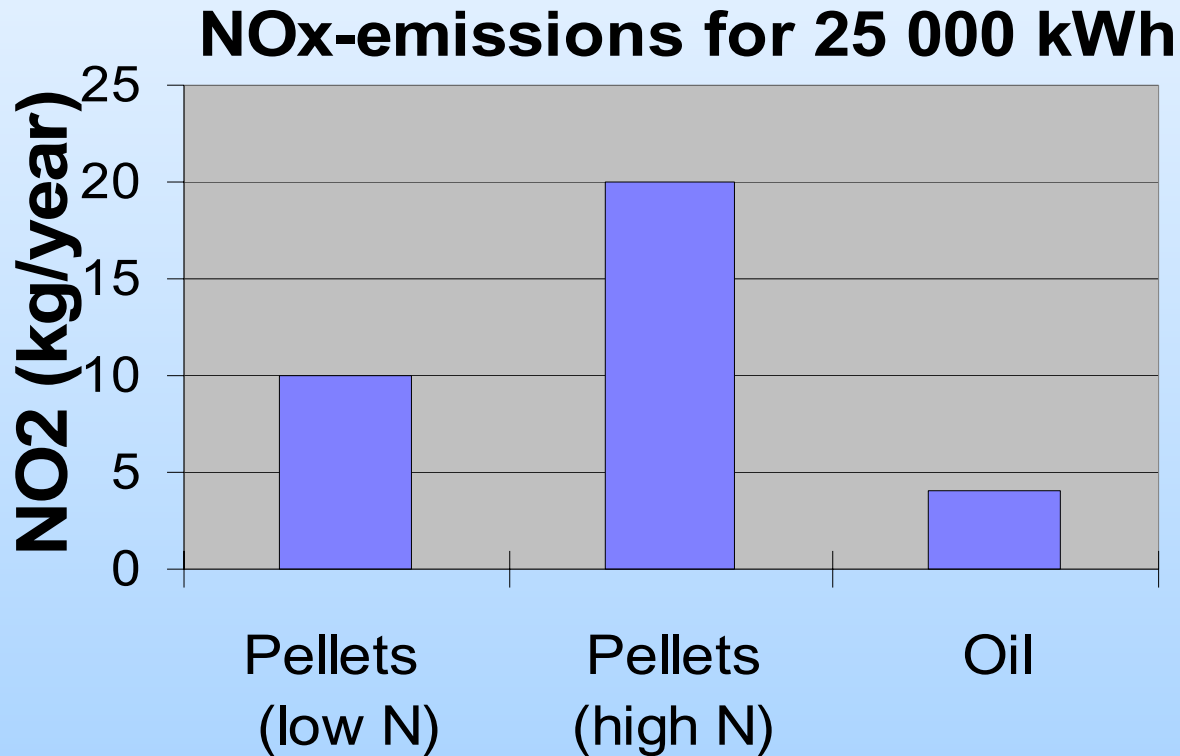


# Fuel-N conversion to NOx is high in pellet burners/stoves

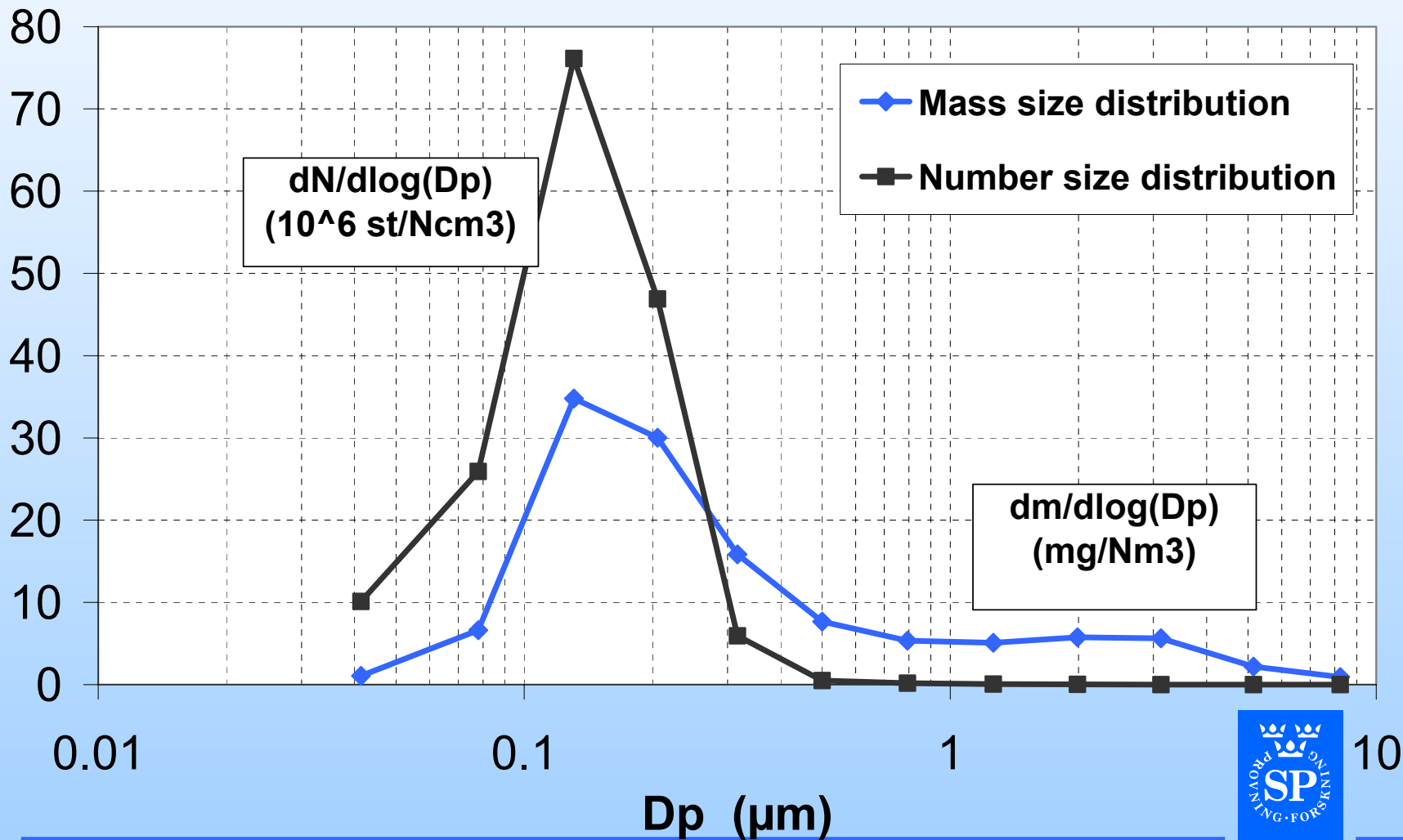


# Oil => Pellets

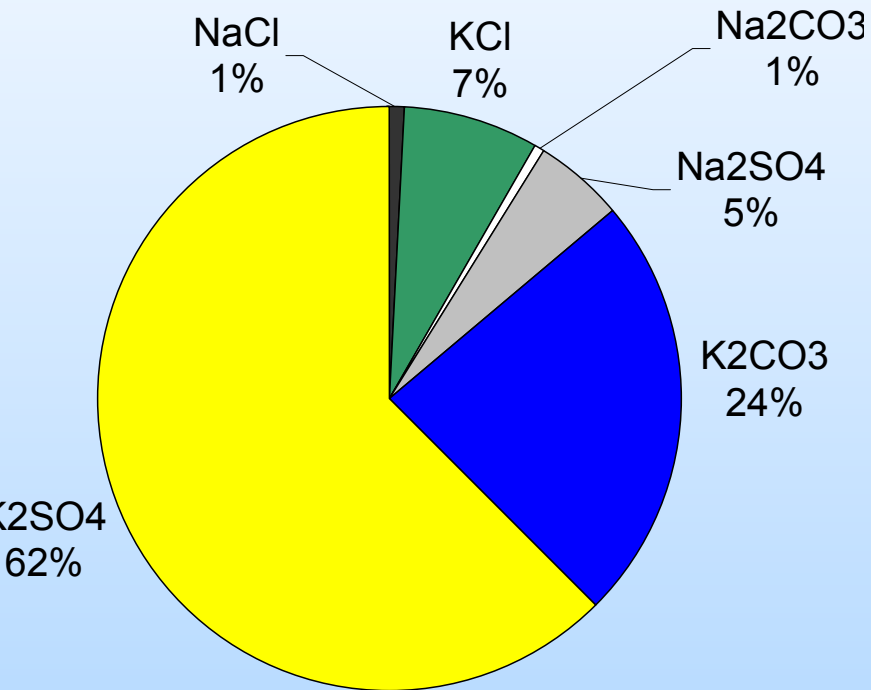
## Higher NOx- emissions



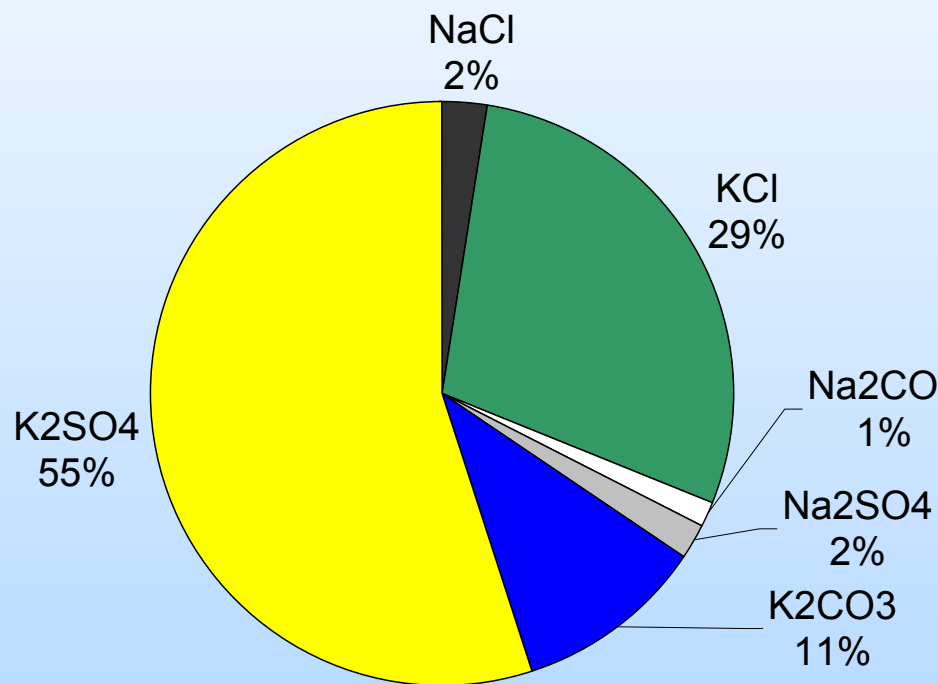
# Grate Boiler, 2,5 MW, Wood Briquettes



# TOF-SIMS Results - Pellet Stove



$D_i = 0.1 \mu\text{m}$



$D_i = 0.3 \mu\text{m}$

# General conclusions

- Steady increase in the use of Bioenergy
- Increasing use of "waste" fuels
- Waste wood
  - Problems with deposit formation/corrosion
    - Fuel quality (Cl, Zn, Pb)
    - Primary/secondary combustion actions
- Increasing use of pellets (small scale)
  - NO<sub>x</sub>
  - Particles